

A Simple and Unambiguous Route to Adamantene: Debromination of 1,2-Dibromoadamantane with Bis(trimethylsilyl)mercury

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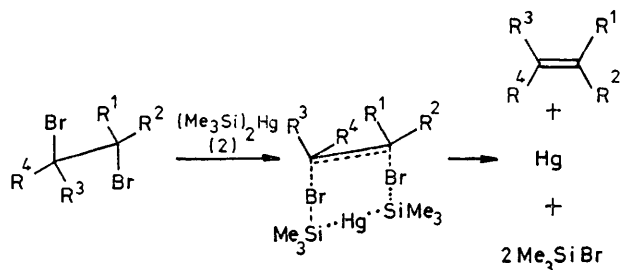
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Summary In a process extensible, in principle, to the preparation of other anti-Bredt alkenes, reaction in boiling benzene of 1,2-dibromo- or 1,2-di-iodo-adamantane with bis(trimethylsilyl)mercury, which dehalogenates 1,2-dibromoalkanes in a concerted, 'four-centre' manner, leads

to the unambiguous generation of adamantene, as shown by trapping by 1,3-diphenylisobenzofuran (30—35% yield of adduct) and 2,5-dimethylfuran (16% yield of adduct).

We report a simple route to adamantene (**1**) which not only leads unambiguously to this elusive species,¹ but is also extensible, in principle, to other anti-Bredt alkenes.

Our route depends on the dehalogenating properties of bis(trimethylsilyl)mercury (**2**)² which is known to debrominate 1,2-dibromoalkanes in a concerted, 'four-centre' manner (Scheme 1).³



SCHEME 1

Thus, reaction of (**2**) (2.5 mol equiv.) with 1,2-dibromoadamantane (1 mol) in the presence of 1,3-diphenylisobenzofuran (5 mol) in boiling benzene after five days gave the hydrocarbon (**3**), 35%, m.p. 204–206 °C.† Taken with the concerted nature of the debromination (*vide supra*), this observation points to the generation of adamantene, its subsequent trapping by the diene, and deoxygenation of the resulting cycloadduct (**4**) to give the product (**3**) (Scheme 2). In support of this it is known⁴ that diphenylisobenzofuran adducts readily lose oxygen under reducing conditions; moreover, in our case, mercury (II) oxide was a by-product.

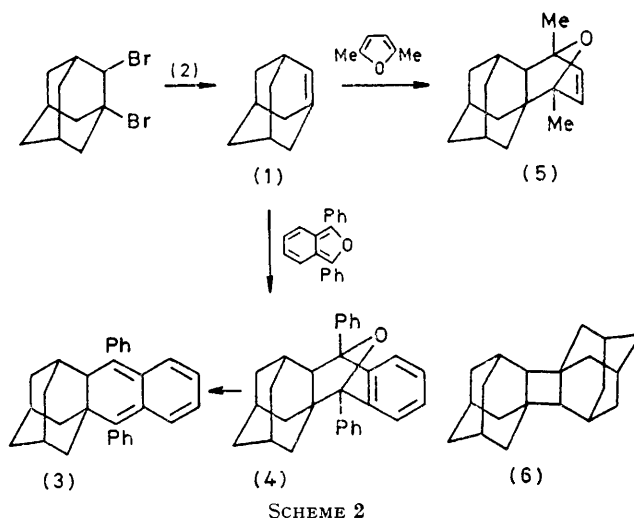
† All new compounds had the expected ¹H and ¹³C n.m.r. and mass spectra and elemental analysis.

¹ (a) J. E. Gano and L. Eizenberg, *J. Amer. Chem. Soc.*, 1973, **95**, 972; (b) D. Lenoir and J. Firl, *Annalen*, 1974, 1467; (c) D. G. Gillespie and B. J. Walker, *Tetrahedron Letters*, 1977, 1673; (d) A. H. Alberts, J. Strating, and H. Wynberg, *ibid.*, 1973, 3047; (e) W. Burns, D. Grant, M. A. McKervey, and G. Step, *J.C.S. Perkin I*, 1976, 234.

² E. Wiberg, O. Stecher, H. J. Andrascheck, L. Kreuzbichler, and E. Staude, *Angew. Chem. Internat. Edn.*, 1963, **2**, 507.

³ S. W. Bennett, C. Eaborn, R. A. Jackson, and R. W. Walsingham, *J. Organometallic Chem.*, 1971, **27**, 195.

⁴ D. E. Applequist, P. A. Gebauer, D. E. Gwynn, and L. H. O'Connor, *J. Amer. Chem. Soc.*, 1972, **94**, 4272.



SCHEME 2

1,2-Di-iodoadamantane behaved similarly, giving the hydrocarbon (**3**) in 30% yield, while similar reaction of 1,2-dibromoadamantane in the presence of 2,5-dimethylfuran as trap gave the 1:1 adduct with adamantene (**5**; 16%), with ¹H n.m.r. spectrum identical with that previously reported.^{1d} In the absence of the above 1,3-dienic traps, 1,2-dibromoadamantane gave dimers of adamantane (*e.g.* **6**) in low yield.

It is noteworthy that previous attempts to trap adamantene of different genesis *via* [2 + 4] cycloaddition with furan and its derivatives have led to adducts in zero to low (<9%) yields.^{1c-e}

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